7. The De Broglie wavelength is calculated from the momentum by

$$\lambda = h / p$$
  
 $\lambda = (6.626 \times 10^{-34} \text{ J s}) / (2.00 \times 10^{-40} \text{ kg m/s})$   
 $\lambda = 3.313 \times 10^{6} \text{ m}$