## Where's The Error?

Careless Carol needs 200 mL of $50 \%$ ethanol, but only has $90 \%$ available. She performs the following calculations, using the formula $\mathrm{V}_{\mathrm{a}}=\underline{V}_{\underline{n}} \frac{\times \mathrm{C}_{n}}{\mathrm{C}_{\mathrm{a}}}$


$$
\begin{aligned}
& V_{n}=200 \mathrm{~mL} \\
& C_{n}=90 \% \\
& C_{a}=50 \% \\
& V_{a}=\frac{200 \mathrm{~mL} \times 90 \%}{50 \%} \\
& V_{a}=360 \mathrm{~mL}
\end{aligned}
$$

Carol realizes it is impossible to make 200 mL of $50 \%$ ethanol by adding water to 360 mL of $90 \%$ ethanol. Where's the error?

